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**Abstract:** We revisit the famous problem of sorting objects using thermodynamics concepts. This helps us to easily derive the well-known time complexity of  $\Omega(n \ln n)$  without any use of the classical computer-science objects (graph, Turing machine, etc.).

**Key-words:** Sort, Boltzmann formula, time complexity

## Sur la complexité temporelle du Tri

**Résumé :** Nous revisitons le problème consistant à trier des objets en utilisant des concepts issus de la thermodynamique. Grâce à ces concepts nous sommes capable de facilement dériver la bien-connue complexité temporelle en  $\Omega(n \ln n)$  sans l'aide des objets classiques de l'informatique (graph, Machine de Turing, etc.).

**Mots-clés :** Tri, Équation de Boltzmann, complexité temporelle

Let be a system with  $n$  objects initially randomly arranged. Using Boltzmann formula, the entropy of a system is:  $S = k \ln(\Omega)$ . Where  $\Omega$  is the number of possible configurations of this system and  $k = 1.381 \cdot 10^{-23} \text{ J.K}^{-1}$  (joule per kelvin) is the Boltzmann constant.

Let us assume we want to sort the  $n$  objects of our system. Then, before sorting the entropy of the system is:

$$S_1 = k \ln(n!) \text{ J.K}^{-1}$$

and after sorting, the entropy is:

$$S_2 = k \ln(m) \text{ J.K}^{-1}$$

Where  $m$  is the number of possible ways to order the objects.

According to thermodynamics, the energy required for a system for going from one state to an other state is  $E = T\Delta S$  where  $T$  is the temperature (in kelvin) where occurs the change of state and  $\Delta S = S_1 - S_2$  is the difference of entropy of the considered states. For sorting our  $n$  objects we therefore need the following energy (in joule):

$$E = Tk (\ln(n!) - \ln(m)) = Tk \ln \left( \frac{n!}{m} \right) \text{ J}$$

Let us suppose we use a machine to make this change of state (find one possible sorting among the  $m$ ). Let us suppose that this machine consumes a power of  $x \text{ W}$  (watt) or  $x \text{ Js}^{-1}$  (joule per second) and works at a constant temperature  $TK$ . Therefore, a lower bound of the time to sort the  $n$  objects is (in seconds):

$$t \geq \frac{T}{x} k \ln \left( \frac{n!}{m} \right) \text{ s}$$

This means that the time complexity of the sort problem is (when  $m$  is assumed to be a constant independent of  $n$ ):

$$\Omega(\ln(n!))$$

As  $n! = \sqrt{2\pi n} \left( \frac{n}{e} \right)^n \left( 1 + O \left( \frac{1}{n} \right) \right)$  (Stirling formula), we obtain the well-known time complexity for the sort problem:

$$\Omega(n \ln n)$$



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